Modulation of electroneutral Na transport in sheep rumen epithelium by luminal ammonia

Khalid Abdoun, Friederike Stumpf, Katarina Wolf, and Holger Martens

Department of Physiology, Faculty of Veterinary Sciences, University of Khartoum, Sudan; and Department of Veterinary Physiology, Free University of Berlin, Berlin, Germany

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Ammonia absorption takes place in the rumen (6), small intestine (41), and the colon (15) and occurs primarily by simple diffusion of the nonionized lipid-soluble NH3 form. However, ionic diffusion (NH4+) has also been demonstrated (5, 8, 12, 15).

In the kidney (31, 52), where NH4+ has to be transported against a considerable concentration gradient from the renal cortex (<0.1 mM) to the medulla and, ultimately, the urine, transport of NH4+ has to be coupled to other ions and replaces other cations in transport proteins such as the NHE3, Na-K-2Cl exchange, K/H exchangers, or NH4+/H+ exchangers belonging to the Rh gene family (31, 52). Conversely, in the rumen and the colon, where NH4+ is in abundant supply, transport via channels is energetically feasible and would explain the increase in short-circuit current observed both in rat colons and rumen epithelium after exposure to this ion (5, 14).

It is generally known that cellular uptake of ammonia affects intracellular pH (pHi) (7, 46). Predominant NH3 uptake tends to alkalize the cytoplasm, whereas the predominant uptake of NH4+ acidifies it. The magnitude and direction of this change in pHi depends on the relative transport rates of NH3 and NH4+, which can be altered by the luminal pH according to the Henderson-Hasselbalch equation. The wide variations in ruminal ammonia concentrations (up to 70 mM) (11) and of pH (5.4–7.4) cause corresponding alterations of NH3 and NH4+ concentrations and flux rates across the rumen epithelium (25). Note that in rumen, NH4+ transport can also be influenced by changes in the potential difference (PD) across the apical membrane (PDa) (5), whereas exchangers are typically not affected by membrane potential (54).

Alteration of pHi is well known to affect cellular transport of Na+ via NHE in various systems (7, 46), including the rumen epithelium (42), and a modulation of Na transport is to be anticipated. In a recent study (1), it was demonstrated that interaction between ammonia and Na+ transport across the isolated rumen epithelium of sheep is modulated by the diet.

The mechanisms underlying this adaptation to an increase in the uptake of nitrogen (urea or protein) are poorly understood. However, it should be noted that luminal application of ammonia induced a significantly higher increase in short-circuit current (Isc) in tissues of concentrate-fed animals than in epithelia of hay-fed animals (1). Therefore, we suggested that the ammonia-induced change in Isc is caused by the uptake of NH4+ in its protonated form. Dissociation of NH4+ within the cytosol should increase the amount of H+ available for exchange with sodium.
Alkalization of ruminal epithelial cells by uptake of ammonia in the form of NH₃ was demonstrated by Müller et al. (42), whereas experiments by Bödeker and colleagues (5, 6) show uptake of NH₄⁺. In these experiments, the ammonia-dependent increase of Iₛₑ was abolished by pretreatment with mucosal quinidine, and the authors suggested involvement of a potassium channel (5, 6). There are conflicting reports on the permeability of the apical membrane of rumen epithelium for potassium, depending on whether luminal potassium was high (20, 35, 50), or low (21, 35). This question obviously needs further clarification.

It was therefore the aim of the present study 1) to determine the flux of ammonia through the rumen epithelium, 2) to change the relative flux rates (NH₄⁺ or NH₃) by variation of luminal pH and PDₐ, 3) to look for effects of altered ammonia flux rates on Na transport, and 4) to characterize the NH₄⁺ conductance.

The patch-clamp data in this study show that ruminal epithelial cells express quinidine-sensitive channels of 130 pS that conduct both potassium and NH₄⁺ but not sodium. Relative permeability of the membrane for potassium rose with concentration. Anomalous mole-fraction effects (29) between K⁺ permeability of the membrane for potassium rose with concentration. Ammonia in the form of NH₃ was demonstrated by Müller et al. (20, 35, 50), or low (21, 35). This question obviously needs further clarification.

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Ussing chamber data demonstrate that variations of luminal pH determine the form of ammonia transport as NH₄⁺ or NH₃ and consequently affect the absorption of Na⁺ depending on the transported form of ammonia. Predominant uptake of NH₄⁺ through potassium channels stimulated and predominant diffusion of NH₃ inhibited Na transport across the rumen epithelium of sheep. Under physiological conditions (pH 6.50 or lower), ammonia stimulates Na transport and may be a modulator of Na absorption via NHE, significantly contributing to the classic effect of short-chain fatty acids (24, 33).

**MATERIALS AND METHODS**

The incubation of rumen epithelium has been described in detail by Martens et al. (39). Briefly, sheep were killed in a local slaughterhouse, and the reticulorumen was removed from the abdominal cavity within 2–3 min. A 250-cm² piece of rumen wall was taken from the ventral sac, cleaned in a buffer solution, stripped from the muscle layer, and taken (some 20 min) to the laboratory in a buffer solution maintained at 38°C. The buffer was gassed with 95% O₂-5% CO₂. Pieces of the epithelium (3 × 3 cm) were mounted between the two halves of an Ussing chamber to give an exposed area of 3.14 cm².

The mounted tissues were bathed on each side with 18 ml buffer solution by using a gas lift system and were gassed with 95% O₂-5% CO₂. The standard electrolyte solution contained in (mM) 90 Na⁺, 5 K⁺, 1 Ca²⁺, 2 Mg²⁺, 25 HCO₃⁻, 5 Cl⁻, 1 H₂PO₄⁻, 2 HPO₄²⁻, 25 acetate, 10 propionate, 5 butyrate, 10 glucose, and 30 tris–HCl (pH 7.4). The standard solution was then gassed with 95% O₂-5% CO₂. The standard solution was then gassed with 95% O₂-5% CO₂.

**Electrical Measurements and Calculation of Na and Cl Flux Rates**

For details, see Martens et al. (38). The transepithelial conductance (Gₑ) was determined by briefly applying a 100-μA current across the tissue in both directions and measuring the resulting change in transmural potential difference (PDₑ), from which Gₑ can be calculated using Ohm’s law. The sum of all electrogenic ions moving across the tissue was determined by measuring the external current (equivalent to Iₑ) necessary for clamping PDₑ to zero.

Fluxes were measured in the short-circuit mode. ²²Na and ³⁶Cl (80 and 100 kBq, respectively) were added to the “hot” side of the epithelium, and three flux periods of 30 min were performed after an equilibration time of 30 min. Samples from the “hot” site were taken before the first and after the last flux period for the calculation of the specific radioactivity. The sample volume was replaced by the corresponding buffer. ²²Na and ³⁶Cl fluxes were determined in separate experiments. Total ammonia flux [mucosal-to-serosal flux (Iₑ) (NH₄⁺ + NH₃)] was calculated from aliquots taken from the serosal side at the beginning and the end of the flux period and determined directly by using an ion analyzer (gas-sensitive electrode; EA 940; Orion).

**Radioactivity**

²²Na and ³⁶Cl were assayed by using a well-type crystal counter (LKB Wallace-Perkin-Elmer, Uberlingen, Germany) and a β-counter (LKB Wallace-Perkin-Elmer, respectively).

**Statistics**

Statistical evaluations were carried out by using the SPSS program version 10.0 for Windows. Results are given as means ± SE. The comparison between the groups was carried out in the form of a repeated-measurement analysis of variance with a two-factorial model without interaction. P values <0.05 were considered significant. N refers to the number of experimental animals, and n refers to the number of tissues.

**Patch-clamp experiments**

Cells from rumen epithelium were cultured on glass coverslips according to established methods (36, 49). Briefly, the stratum corneum (26) was removed from ruminal papillae by fractional trypsinisation, and the last (fourth or fifth) fraction was plated out in cell culture dishes. The number of cornified cells in the culture increased rapidly after seeding. Coverslips were removed from the culture dishes for experiments 3–8 days after seeding or, alternately, 1–5 days after reseeding from primary culture. Previous studies in our lab suggest that cells at this stage have differentiated and express proteins that are found in the apical membrane of ruminal epithelium (36, 48).

Immediately before use, cells were gently cleansed with trypsin (0.02%, Biochrome, Berlin, Germany) for 1 min and introduced into a perfusion chamber on the stage of an inverted microscope where they were superfused by warmed Ringer solution (37.5°C; TC01 and PH01, Lohmann Research Equipment, Castrop-Rauxel, Germany). All patch-clamp experiments were performed essentially as in a previous study (36). Only cells in which washout occurred, demonstrating seal stability, were used for evaluation. Capacitance was measured regularly; cells in which swelling occurred were excluded. Note that usually the seals of such cells ruptured quickly, and stable measurements with washout could not be obtained.

Currents were recorded using an EPC 9 patch-clamp amplifier (HEKA Elektronik, Lambrecht, Germany). Pulse generation, data collection, and analysis was performed using TIDA software (HEKA Elektronik) and filtered with a 2.9-kHz Bessel filter. Records were corrected for capacitance. Positive ions flowing into the pipette correspond to a negative current and are depicted in figures as going downward. For inside-out patches, the pipette potential corresponds to the negative membrane potential.

Two types of pulse protocols were used. Either current response was recorded at 100 Hz using a protocol that generated steps of 200-ms duration to voltages between -120 and 100 mV in 20-mV steps, returning to a holding potential of -40 mV for 200 ms in between [“pulse protocol I” (36)]. This protocol was repeated continuously to allow the monitoring of current responses of the cells to...
changes in external solution. In addition, conventional voltage pulse protocols were used that recorded data at a much higher sampling rate of 5 kHz ["pulse protocol II" (36)]. As before, holding potential was −40 mV, and voltages ranged from −120 to 100 mV, but the step size was 10 mV. For single-channel experiments, data were sampled at 10 kHz.

**Solutions and Chemicals (Patch Clamp)**

Pipette solution for whole cell experiments, designated as “K-gluconate” contained (in mM) 1 KH2PO4, 10 HEPES, 0.8 CaCl2, 0.9 MgSO4, 5 EGTA, 123 K-gluconate, and 10 NaCl. Extracellular NaCl solution contained the following ion concentrations (in mM): 130 NaCl, 1 NaH2PO4, 5 KCl, 10 HEPES, 1.7 CaCl2, and 0.9 MgCl2. With the use of this basic recipe, NaCl was substituted by either choline chloride or NH4Cl or KCl in the solutions designated by these ions. In solutions labeled “NH4” (65 mM), “K” (65 mM), or “NH4-chloride” contained ion compositions were equal in composition but contained the respective ions.

For inside-out single-channel experiments, standard 130 mM NH4Cl solution was used to fill the pipette (facing the external side of the membrane), whereas calcium was reduced in the KCl bath solution facing the cytosolic side and contained (in mM) 130 KCl, 3 NaCl, 5 K-gluconate, and K-gluconate were equal in composition but contained the respective ions instead of KCl. Liquid junction potentials were corrected according to established methods (3).

**Analysis**

To compare whole cell data from different cells with each other, initial inward and outward current in NaCl solution at −120 and 100 mV were assigned the value of 100%. All other currents were seen in relationship to these values. Significance testing was performed using the paired Student’s t-test and standard software. The number n refers to the number cells used; care was taken to include cells from different animals.

Reversal potentials were estimated by linear regression between the zero-current values just above and just below the zero level for each new steady state of $I_{sc}$ was obtained after 10–15 min, and the rise of $I_{sc}$ ($\Delta I_{sc}$) exhibited saturation kinetic properties (Fig. 1; pH 7.4).

This change of $I_{sc}$ was pH dependent and was related to the luminal ammonia concentration. At pH 6.4, effects of an influx of NH3 should be lowest. However, $I_{sc}$ changed most at this pH, suggesting influx of ammonium in the ionized form.

**Determination of Ammonia Flux Rates**

It is impossible to distinguish between transport of NH3+ through a potassium channel and of Na+ via the electrogenic pathway (34, 36) by measuring $I_{sc}$ alone. In addition, no information on the transport of ammonia as NH3 is obtained. Therefore, ammonia flux rates were measured directly.

Total ammonia was kept constant (30 mM), and concentration of NH3+ varied by increasing the pH from 6.4 to 6.9 and 7.4. This should induce an increasing gradient in NH3+ concentration at an almost constant concentration of NH3+. At pH 6.4, a steady-state ammonia flux [Jms (NH3 + NH3+)] of 0.82 μmol·cm−2·h−1 could be measured from the mucosal to the serosal side. An increase in pH on the luminal side of the membrane resulted in a significant (P < 0.05) stimulation of ammonia flux rates (Fig. 2).

The intercept (y-axis, NH3 = 0) represents the flux of NH3+ and shows that it has the same magnitude (0.7 μeq·cm−2·h−1) as the ammonia-dependent increase of $I_{sc}$ at pH 6.4 (Fig. 1). It thus appears that at a pH of 6.4, ammonia is predominantly transported as NH3+ across the apical membrane and should be seen as the major ion responsible for the increase in $I_{sc}$. At pH 6.9, total flux is doubled and the flux rates of NH3 and NH3+ are almost equal. The total flux of ammonia is 3.41 ± 0.53 μeq·cm−2·h−1 at pH 7.4, and hence, NH3 transport clearly exceeds that of NH3+. This change in ammonia transport should influence pH2 (42) and, possibly, Na transport via NHE.

**Effect of Ammonia on Na and Cl Transport at pH 7.4**

Increasing luminal ammonia concentrations in a CO2/HCO3−-free buffer at a luminal pH of 7.4 significantly (P < 0.05) decreased the mucosal-to-serosal flux of sodium [Jms (Na)].

**RESULTS**

**Effect of Increasing Luminal Ammonia Concentrations on $I_{sc}$**

In a first series of experiments, the effect of adding different concentrations of ammonia to the luminal side of rumen epithelium was studied using the Ussing chamber technique. At the end of the equilibration period, the $I_{sc}$ and the total tissue $G_{I}$ exhibited constant values of 0.68–0.80 μeq·cm−2·h−1 and 2.13–2.68 mS/cm2, respectively, reflecting elective sodium transport (34, 36). This did not change significantly during the next 30 min in control tissues. As in the study by Bödeker and Kernen (5), luminal ammonia caused a significant increase of $I_{sc}$ up to 0.45 μeq·cm−2·h−1 (Fig. 1).
In contrast, ammonia significantly \((P < 0.05)\) stimulated Na transport at pH 6.4 (Table 4). \(J_{\text{ms}}(\text{Na})\) and \(J_{\text{net}}(\text{Na})\) increased from 6.01 ± 0.42 and 4.37 ± 0.42 to 7.90 ± 0.47 and 6.50 ± 0.50 \(\mu\text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1}\), respectively. Luminal ammonia (30 mM) did not show any significant effect on Cl flux rates at pH 6.40 (Table 4).

Note the different responses of \(I_{sc}\) and \(G_t\) to ammonia. To measure \(I_{sc}\), PDt was clamped to zero, and an activation of voltage-dependent currents appears unlikely. \(I_{sc}\) showed an almost linear correlation with the concentration of ammonia at pH 6.4 (where influx of NH\(_3\) is minimal; Table 4 and Fig. 1). Conversely, the nonlinear increase of \(G_t\) with the concentration of ammonia may reflect activation of voltage-dependent currents (34, 36).

**Effect of PDt**

Because NH\(_4^+\) is charged, uptake across the luminal membrane through a K channel should be modulated by the PDt. PDt was altered by variation of PDt (34), and the effect on the transport of Na both in the presence and the absence of luminal ammonia (30 mM) was studied. All solutions were titrated to a pH of 6.4, so that ammonia should be present almost exclusively in the ionized form.

Imposing a PDt of +25 mV (serosal side positive) significantly depolarizes PDt by 15 mV from some ~50 to ~35 mV (34). This PDt is too low to induce significant stimulation of electrogenic sodium transport (34) against a high background of electroneutral sodium transport (stimulated by presence of short-chain fatty acids in the current study).

In line with this, a PDt of +25 mV only led to a small, nonsignificant reduction of \(J_{\text{ms}}(\text{Na})\) and \(J_{\text{net}}(\text{Na})\) (Table 5). Likewise, serosal-to-mucosal flux \(J_{\text{sm}}(\text{Na})\) (Na) was not significantly changed either by changing PDt or by the addition of ammonia. All changes probably represent effects on paracellular, passive flow.

As before, ammonia (30 mM) stimulated \(J_{\text{ms}}(\text{Na})\) and \(J_{\text{net}}(\text{Na})\), both under short-circuit conditions and at +25 mV \((P < 0.05\) vs. control).

However, in the presence of ammonia (30 mM), \(J_{\text{ms}}(\text{Na})\) and \(J_{\text{net}}(\text{Na})\) could be reduced significantly by increasing PDt from 0 to +25 mV \((P < 0.05)\). The potential-induced decrease in \(J_{\text{net}}(\text{Na})\) \(J_{\text{net}}(\text{Na})\) at PDt 0 mV \(J_{\text{net}}(\text{Na})\) at PDt +25 mV was significantly greater in the presence of ammonia \((1.59 \pm 0.27 \mu\text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1})\) than in its absence \((0.77 \pm 0.28 \mu\text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1})\). This suggests that depolarization of the apical membrane reduced the influx of NH\(_4^+\) needed to stimulate the NHE in the presence of ammonia.

**Table 1. Effect of increasing luminal ammonia concentrations on Na flux rates, \(I_{sc}\), and \(G_t\) at luminal pH 7.4 in the absence of CO\(_2\)/HCO\(_3^-\) (HEPES buffer)**

<table>
<thead>
<tr>
<th>NH(_4^+)Cl, mM</th>
<th>(J_{\text{ms}})</th>
<th>(J_{\text{net}})</th>
<th>(I_{sc})</th>
<th>(G_t)</th>
<th>(N/n)</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>2.84±0.09</td>
<td>0.82±0.08</td>
<td>2.01±0.11</td>
<td>0.61±0.13</td>
<td>1.99±0.12</td>
</tr>
<tr>
<td>5</td>
<td>2.17±0.01*</td>
<td>0.77±0.04</td>
<td>1.40±0.09*</td>
<td>0.81±0.05</td>
<td>2.00±0.09</td>
</tr>
<tr>
<td>15</td>
<td>1.63±0.11*</td>
<td>0.64±0.09</td>
<td>0.99±0.06*</td>
<td>0.87±0.07</td>
<td>2.01±0.22</td>
</tr>
<tr>
<td>30</td>
<td>1.43±0.19*</td>
<td>0.69±0.10</td>
<td>0.74±0.12*</td>
<td>1.02±0.11*</td>
<td>2.24±0.32*</td>
</tr>
</tbody>
</table>

Values are means ± SE; \(N\) is no. of animals; \(n\) is no. of tissues. *\(P < 0.05\) for comparison with the control group (0 mM NH\(_4^+\)Cl); Na flux rates, Cl flux rates, and short circuit current \(I_{sc}\) \((\mu\text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1})\); \(G_t\), transcellular conductance \((\text{mS/cm}^2)\); \(J_{\text{ms}}\), mucosal-to-serosal flux; \(J_{\text{sm}}\), serosal-to-mucosal flux; \(J_{\text{net}}\), net flux.
Effect of Amiloride

The assumption that the stimulatory effect of ammonia (30 mM) on Na transport at pH 6.4 is caused by enhanced NHE activity was tested by blocking this exchanger by 1 mM mucosal amiloride (38), completely blocking the effects of ammonia on net Na transport (Table 6). Note that amiloride had no effect on \( I_{sc} \).

Effect of Mucosal K

To test for mutual inhibition of \( \text{NH}_4^+ \) and \( K^+ \) in a common pathway, we studied the modulation of Na transport by K. Luminal solutions contained 15 mM ammonia at pH 6.4. Increasing luminal K concentration from 5 (control) to 45 mM caused a significant reduction of \( J_{\text{net}}(\text{Na}) \) from 8.70 ± 0.67 (control) to 6.01 ± 0.65 \( \mu \text{eq cm}^{-2} \text{h}^{-1} \), respectively, with concurrent significant reduction of \( J_{\text{net}}(\text{Na}) \). \( J_{\text{sm}}(\text{Na}) \) remained unchanged (Table 7).

Patch-clamp Experiments

Patch-clamp experiments in the whole cell configuration. K-Gluconate pipette solution: effect of \( \text{NH}_4\text{Cl} \). In a first series of experiments, rumen epithelial cells were brought into the whole cell configuration using the K-gluconate solution described in the MATERIALS AND METHODS (Fig. 4).

Cells were superfused with high-sodium (NaCl) solution until current had stabilized, indicating replacement of the cytosolic fluid with the pipette solution. Mean inward current density at −120 mV was −12 ± 3 pA/PF, mean outward current density at +100 mV was 32 ± 8 pA/PF (n = 33), in good agreement with a previous study (36).

When NaCl was replaced by \( \text{NH}_4\text{Cl} \) (130 mM) in the external solution, inward current rose to 189 ± 17% of the original value (100%) measured in NaCl solution at a pipette potential of −120 mV (\( P < 0.01, n = 17 \)). Outward current at +100 mV rose to a mean level of 132 ± 10% (\( P = 0.007, n = 17 \); Fig. 4A). After washout of \( \text{NH}_4\text{Cl} \) with NaCl, inward and outward currents recovered to 112 ± 12 and 102 ± 15%, respectively, of the original values in NaCl solution (\( P = 0.3 \) and \( P = 0.9 \), no significant difference to the value before application of ammonia). Reversal potential rose in all cells studied from a mean value of −33 ± 4 (NaCl) to −6 ± 4 mM (\( \text{NH}_4\text{Cl} \); \( n = 17, P < 0.001 \)) and back to −25 ± 3 mM (NaCl; \( P < 0.01 \)).

Conversely, replacement of NaCl in the bath solution with choline chloride did not induce significant changes in either inward current (85 ± 12%, \( n = 7, P = 0.2 \)), outward current (88 ± 9%, \( P = 0.9 \)), or reversal potential level (−25 ± 7 mV, \( P = 0.7 \)), in line with a previous study (36).

Exposure to \( \text{BaCl}_2 \) (5 mM), known to block potassium channels, resulted in a significant decrease in \( \text{NH}_4\text{Cl} \)-induced inward current to 70 ± 9% (\( n = 7, P < 0.05, -120 \text{ mV} \)) of the level in \( \text{NH}_4\text{Cl} \) solution without \( \text{BaCl}_2 \) (100%), whereas outward current dropped to 77 ± 7% (\( n = 7, P < 0.01, +100 \text{ mV} \)) of the original outward current level. Conversely, TEACl (10 mM) had no significant effect on the inward current induced by \( \text{NH}_4\text{Cl} \) (102 ± 20%, \( n = 4, P = 0.9 \)), whereas outward current dropped slightly to 80 ± 7% (\( n = 4, P < 0.05 \)). Reversal potential was not significantly altered by either the addition of \( \text{BaCl}_2 \) (3 ± 8 mV, \( n = 7, P = 0.1 \)) or TEA (−10.3 ± 15 mV, \( n = 4, P = 0.05 \)) to NaCl bath solution.

K-Gluconate pipette solution: effect of KCl. Similar effects were observed when NaCl bath solution was replaced with KCl bath solution (Fig. 4B). This induced a rise of both inward current and outward current to 344 ± 66% (\( n = 8, P = 0.007, -120 \text{ mV} \)) and 218 ± 80% (\( n = 8, P = 0.003, +100 \text{ mV} \)), respectively, with reversal potential rising to −10 ± 7 mV (\( n = 8, P = 0.01 \)).

Table 2. Effect of increasing luminal ammonia concentrations on Cl flux rates, \( I_{sc} \) and, \( G_{i} \) at luminal pH 7.4 (\( \text{CO}_2/\text{HCO}_3^- \) buffer)

<table>
<thead>
<tr>
<th>CI</th>
<th>( J_{\text{in}} )</th>
<th>( J_{\text{out}} )</th>
<th>( J_{\text{net}} )</th>
<th>( I_{sc} )</th>
<th>( G_{i} )</th>
<th>( N/m )</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>3.76±0.60</td>
<td>2.37±0.35</td>
<td>1.39±0.33</td>
<td>1.15±0.08</td>
<td>3.39±0.34</td>
<td>4/8</td>
</tr>
<tr>
<td>30</td>
<td>5.31±0.29</td>
<td>2.22±0.32</td>
<td>3.09±0.30*</td>
<td>1.59±0.09*</td>
<td>3.72±0.29</td>
<td>4/8</td>
</tr>
</tbody>
</table>

Values are means ± SE. For corresponding data on Na fluxes, see Ref. 1. *\( P < 0.05 \).
In KCl solution, TEACl (10 mM) blocked KCl\(^{-}\)-induced inward and outward current to values of 58 ± 20 (n = 4, P < 0.05, -120 mV) and 66 ± 18% (n = 4, P < 0.01, +100 mV), respectively. Effects of TEACl on cells in NaCl solution did not reach significance level (n = 3). The effects of BaCl\(_2\) (5 mM) were less pronounced, with inward current dropping to 72 ± 17% (n = 4, P < 0.05) of the original level in the presence of KCl, whereas changes in outward current did not reach significance level [92 ± 7% (n = 4, P < 0.05)]. The currents were also blocked by quinidine (100 μM) with inward current sinking to 82 ± 3% (n = 3, P < 0.05), outward current to 47 ± 23% (n = 3, P < 0.05), whereas reversal potential remained the same as in KCl solution.

K-GLUCONATE PIPETTE SOLUTION: INTERACTION OF KCl AND NH\(_4\)Cl AT pH 6.4. Ussing chamber experiments suggest that influx of NH\(_4\)\(^{+}\) into ruminal epithelial cells is inhibited by application of K\(^{+}\). To test for the hypothesis that K\(^{+}\) and NH\(_4\)\(^{+}\) use a common pathway, cells were exposed to solutions containing either 65 mM NH\(_4\)Cl, 65 mM KCl, or a mix of both ions (osmolarity adjusted with choline chloride). All solutions were titrated to a pH of 6.4, at which the concentration of NH\(_3\) should be minimal (Fig. 4C).

Exchanging pH of external NaCl solution from 7.4 to 6.4 did not significantly alter either inward current, outward current, or reversal potential (-33 ± 2 mV, n = 8, P = 0.1). Application of NH\(_4\)Cl (65 mM) increased inward current (-120 mV) from 100 to 167 ± 19% (n = 6, P = 0.01), whereas changes in outward current did not reach significance (116 ± 8%, P = 0.1). Reversal potential rose from -33 ± 2 to -24 ± 2 mV (P = 0.02), significantly less than that obtained with 130 mM of NH\(_4\)Cl [P (unpaired) = 0.009]. Conversely, currents at -120 mV were not significantly different from those observed in solutions containing 130 mM NH\(_4\)Cl (Fig. 4C), suggesting effects of saturation.

The application of KCl (65 mM) increased mean inward current to 180 ± 21% [n = 4, P (paired) = 0.03], whereas outward current and reversal potential only rose slightly to 137 ± 16% (P = 0.2) and -26 ± 4 mV (P = 0.1), respectively. These values are not significantly different from those obtained with NH\(_4\)Cl (65 mM) solution.

If the independence principle holds (29), application of a mix of NH\(_4\)Cl (65 mM) and KCl (65 mM) should result in additive depolarization. Indeed, after application of a mix of both salts, cells depolarized to an end value of -18 ± 2 mV (P = 0.004 vs. NaCl) in KCl + NH\(_4\)Cl solution or by a difference of 15 ± 9 mV. This value corresponds roughly to the numeric sum of the depolarizations obtained with NH\(_4\)Cl (65 mM) and KCl (65 mM; 12 ± 5 mV, P = 0.3) as calculated for the individual cells. However, the current responses to NH\(_4\)Cl + KCl solution were much lower than those predicted by the independence principle. Inward current rose relative to NaCl, reaching 175 ± 27% (n = 7, P = 0.03). Relative to NH\(_4\)Cl (65 mM), inward current dropped in three cells in KCl + NH\(_4\)Cl solution (Fig. 4C), rose slightly in two others, and remained unaltered in one cell, with no significant net effect on the mean value. Independence theory would have predicted a rise in all cells by ~80%. A rise of this magnitude was not observed in any of the cells.

NH\(_4\)Cl + KCl solution did not induce significant changes in outward current level vs. NaCl solution (112 ± 11%, P = 0.3). However, a decrease versus the level in NH\(_4\)Cl (65 mM) solution could be observed, with outward current dropping visibly in five of six cells (to 85 ± 5% of the level in NH\(_4\)Cl (100%; n = 5, P = 0.04; Fig. 4C).

Further reduction of NH\(_4\)Cl concentration to 32.5 mM (choline replacement) resulted in a smaller, but still clearly visible, induction of inward current (n = 2).
**Choline chloride pipette solution.** To minimize interference between K⁺ and NH₄⁺, cells were filled with choline chloride pipette solution (Fig. 5, A and B).

In these cells, reversal potential was 11 ± 3 mV (n = 7) in NaCl bath solution, significantly higher than in the potassium gluconate-filled cells (P < 0.01). As before, no changes in inward or outward current could be detected when NaCl bath solution was replaced with choline chloride bath solution (100 ± 5%, P = 0.9, n = 8, −120 mV and 100 ± 7%, P = 0.9, n = 8, +100 mV; Fig. 5A). Reversal potential was not significantly different than in the presence of external sodium (5 ± 1 mV, P = 0.09). Replacement of NaCl external solution with NH₄Cl solution resulted in an increase of inward current to 221 ± 36% (n = 7, P < 0.01), whereas changes in mean outward current were not significant (140 ± 36%, n = 7, P = 0.2; Fig. 5A). Reversal potential in NH₄Cl bath solution [19 ± 2 mV (n = 12)] was significantly higher than in either NaCl (P < 0.01) or choline chloride solution (P < 0.01). Iberiotoxin (10⁻⁷ M), a blocker of high-conductance calcium-activated potassium channels, had no impact on either the current induced by NH₄Cl (100 ± 8%, P = 0.9, n = 4 and 86 ± 16%, P = 0.4, n = 4) or the reversal potential (21 ± 3 mV, P = 0.6).

**NH₄Cl pipette solution.** Cells were filled with an NH₄Cl pipette solution. In NaCl bath solution, reversal potential was −14 ± 4 mV (n = 12), significantly lower than in choline chloride-filled cells (P = 0.007), reflecting efflux of NH₄⁺.

When cells were superfused with NH₄Cl solution (pH 5.8, reduction of NH₃), cells depolarized to 4.1 ± 3 mV (n = 8, P = 0.01) with inward current level at 404 ± 139% of the value in NaCl solution, (n = 8, P < 0.05, −120 mV) and outward current at 213 ± 66% (n = 8, P < 0.01, +100 mV; Fig. 5, C and D). Addition of quinidine (100 μM) to NH₄Cl solution lowered inward current to 42 ± 17% of the value in NH₄Cl solution (n = 3, P < 0.05) and outward current to 43 ± 3% (n = 3, P < 0.01). Reversal potential remained the same (5 ± 4 mV, P = 0.2). Values recovered to 90 ± 20% (n = 3, P = 0.6, inward) and 85 ± 4% (n = 3, P = 0.06, outward) of the level at the beginning of the experiment after washout with NaCl buffer. Superfusion with KCl raised both inward and outward current, both of which were blocked by quinidine (n = 1). Reversal potential changed from −6 (NaCl) to 26 mV (KCl) and to 6 mV (KCl + quinidine).

At a concentration of 65 mM NH₄Cl, inward current of the NH₄Cl-filled cells rose to 197 ± 54% (n = 4, pH = 6.4), reversal potential to 2.5 ± 6. Inward current did not change significantly when KCl was given in addition to NH₄Cl (each at 65 mM; 205 ± 52%, n = 4), whereas reversal potential rose to 19 ± 4 mV [n = 4, P < 0.01 vs. NH₄Cl (65 mM) alone]. KCl (65 mM) hyperpolarized two of three tested cells, suggesting that in these cells, stimulation of NH₄⁺ efflux (at 130 mM) exceeded the additional influx of K⁺ (65 mM).

**Relative permeability ratios.** In a previous study (36), we were able to show that K-glucanate-filled cells were depolarized by removal of external chloride, so that in NaCl solution, P_K/P_Na must be <1.8 ± 0.5. From these measurements, it appears that the larger part of the outward current and the reversal potential in NaCl solution are due to chloride influx and not to K efflux. A similar estimate of <1.6 can be obtained (indirectly) by comparing cells filled with Cs-methanesulfonate, CsCl, choline chloride, and K-glucanate (36).

On the other hand, superfusion of the cells with KCl solution also resulted in a depolarization of similar magnitude as that observed after removal of Cl. If P_K/P_Na is calculated from the reversal potentials (as described in MATERIALS AND METHODS) in NaCl solution and in KCl solution (both 130 mM), a much higher value of P_K/P_Na = 5 ± 2 is obtained. This discrepancy can be resolved if one assumes that permeability for potassium rises with the concentration of this ion in the bath solution. This assumption is confirmed by observing the increase in (TEACl sensitive) outward current after superfusion with KCl solution.

Other permeability ratios at 130 mM were: P_Na/P_choline = 1.0 ± 0.1, P_Na/P_K = 2.8 ± 0.4, and P_H2N/P_K = 0.7 ± 0.4. A relative permeability ratio for NH₄⁺ efflux in NaCl solution can be calculated from NH₄Cl-filled cells superfused with NaCl solution and choline-filled cells superfused with NaCl solution, yielding P_Na/P_choline = 2.6 (130 mM), which is in good agreement with the value obtained for P_H2N/P_Na (130 mM).
mM) from K-gluconate-filled cells. This suggests that the permeability of ruminal cells for NH₄⁺/H⁺ is not as dependent on concentration as that of K⁺/H⁺.

**Patch-clamp experiments in the single-channel configuration.** Experiments in the inside-out configuration of the patch-clamp technique with symmetrical 130 mM NH₄Cl solution in pipette and bath showed a channel with a conductance level of 135 pS (n = 5; Fig. 6, A and D). When NH₄Cl was replaced by NaCl, downward channel openings corresponding to inward current into the pipette at negative potentials disappeared; outward openings were rare and too small for further evaluation (Fig. 6B). Conversely, when NH₄Cl was replaced with an equimolar amount of KCl, channel openings reappeared and could also be observed when KCl was replaced by potassium gluconate in the bath solution (Fig. 6C). Removal of ATP (1 mM) from the cytosolic side (n = 7) did not affect channel openings. The data were fitted by the Goldman-Hodgkin-Katz equation for the two ions of sodium and potassium (Fig. 6E), yielding a permeability ratio of Pₐₙ₄/Pₜₙ₃ = 0.5 ± 0.2 (n = 7). Note the difference between permeability ratio as derived from the reversal potential and the ratio of absolute current values at negative potential levels (29). Such deviations from the independence principle are typically observed when ions interact with each other in the long, narrow pore of a potassium channel (29).

**DISCUSSION**

Very little is known about the interaction between ammonium transport and the NHE in transporting epithelia (1, 13, 14). To our knowledge, the facilitation of sodium absorption by acidification of luminal pH in the presence of ammonia has not been reported before. This is surprising in light of the high concentrations of ammonia to be found both in the rumen of sheep and cattle (25) and the human gut (55). Note that the absorption of fatty acids (short-chain fatty acids) is linked to acidification and thus to the uptake of sodium via the sodium-proton exchanger (NHE) in both tissues (24, 33).

In diarrhea, pH values of stool water >8.00 have been measured (44) with a concurrent rise in NH₃ concentration. This should elevate intracellular pH and reduce the number of

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Fig. 4. A: current measured in ruminal epithelial cell filled with K-gluconate pipette solution in 130 mM NaCl solution (i), in 130 mM NH₄Cl solution (ii), and after washout with NaCl solution (iii), all using voltage protocol II [see MATERIALS AND METHODS and (36)]. Not only inward, but also outward current was stimulated. B: current measured in ruminal epithelial cell filled with K-gluconate pipette solution in response to voltage protocol I. Both inward and outward current increased in response to the application of KC1 solution. Both the NH₄Cl and the KCl-induced currents could be blocked by quinidine (100 μM). C: anomalous mole fraction effects were observed when cells were exposed to a mix of KCl and NH₄Cl. In the bath, NaCl was replaced by NH₄Cl (at 65 or 130 mM) or a mix of KCl and NH₄Cl (at 65 mM each), with osmolarity adjusted in all solutions (choline chloride). pH of all solutions was titrated to 6.4. D: mean values of currents from all cells filled with K-gluconate solution, normalized to inward current in NaCl solution at the beginning of the experiment at a pipette potential of −120 mV (−100%). Note induction not only of inward current, but also of outward current and the nonadditive responses to a mix of ions (see DISCUSSION).
protons available for extrusion by the NHE. This mechanism could help to explain the loss of sodium and water in these conditions.

In sheep and cattle, animals not given sufficient time to adapt to a diet rich in energy and protein (as in fattening) are prone to a condition known as ruminal acidosis (22, 32). Large shifts in the \( \text{NH}_3/\text{NH}_4^+ \) ratio occur, and uptake of protons not only due to the uptake of fatty acids as previously described (24) but also as \( \text{NH}_4^+ \), should be discussed. Thus a better understanding of the uptake routes for ammonia in the rumen and the interaction between ammonia and the NHE appears necessary.

The current study demonstrates that ammonia crosses the ruminal epithelium not only in the form of lipophilic \( \text{NH}_3 \) but also in its ionized form as \( \text{NH}_4^+ \). The relative transport rates of the two forms depend on ruminal pH, with an impact on the absorption rate of Na\(^+\) by the rumen epithelium via NHE.

The suggestion of \( \text{NH}_4^+ \) uptake at pH 6.40 through a putative, quinidine-sensitive K channel by Bödeker and colleagues (5, 6) is supported by the present study. In patch-clamp experiments, we were able to show that ruminal epithelial cells express quinidine-sensitive channels of 130 pS that conduct potassium and \( \text{NH}_4^+ \), but not sodium, with interference when K\(^+\) and \( \text{NH}_4^+ \) are applied simultaneously. Competition between \( \text{NH}_4^+ \) and K\(^+\) for a common, electrogenic uptake pathway could also be demonstrated in Ussing chamber experiments.

The pH-dependent absorption of ammonia from the rumen has long been known (25). The increase of ammonia toxicity at a luminal pH greater than 7.30 (16) supports the assumption of an enhanced absorption of \( \text{NH}_3 \) with increasing pH. This suggestion is supported by the determination of ammonia flux rates in this study. The wide physiological variation of ruminal pH (5.4–7.4) and ammonia concentrations (up to 70 mM) only has a minimal effect of 1.8% on the concentration of \( \text{NH}_3 \) but great impact on free \( \text{NH}_3 \), which increases from 0.07 to 0.68 mM (at 30 mM luminal ammonia), resulting in an almost linear increase in \( \text{NH}_3 \) flux through the epithelial cells (Fig. 2). Conversely, at the pH values below 6.9, which predominate in the rumen, uptake of \( \text{NH}_4^+ \) as a source of protons must be considered. It is a well-known fact that both \( \text{NH}_3 \) and \( \text{NH}_4^+ \) affect pHi in opposite directions (7, 42, 46) and should thus have an impact on the amount of Na\(^+\) transported by the NHE.

Recent in vitro studies with isolated rumen epithelial cells (REC) have demonstrated that recovery from acidification of pHi is reduced by inhibitors of NHE and that exposure of REC to 20 mM NH\(_4\)Cl induces an increase in pHi to 7.8 (42), which inhibits NHE (2). In line with this observation, our findings indicate that in the intact rumen epithelium, electroneutral Na transport is abolished at pH 7.4. The remaining \( J_{\text{net}} \) (Na) at 30 mM ammonia represents electrogenic Na transport and accounts for \( I_{\text{Na}} \), \( J_{\text{net}} \) (Na) and \( I_{\text{Na}} \) have the same magnitude if the measured \( I_{\text{Na}} \) (1.02 \( \mu \text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1} \)) is corrected for the \( \text{NH}_4^+ \)-dependent component (some 0.4 \( \mu \text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1} \)) of \( I_{\text{Na}} \) (1.02–0.40 = 0.62 \( \mu \text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1} \)) vs. a \( J_{\text{net}} \) (Na) of 0.74 \( \mu \text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1} \); see Table 1).

Thus it emerges that at a ruminal pH of 7.4 and at physiological ammonia concentrations, electroneutral Na transport is inhibited (Fig. 7A). Lowering the luminal pH from 7.4 to 6.9 does not alter Na fluxes. A further decrease of luminal pH to 6.4 causes a concentration-dependent increase in Na transport (Fig. 7B). The involvement of the NHE in ammonia induced enhanced Na transport is supported by the observation that mucosal amiloride (1 mM) prevents the stimulating effect of ammonia at a luminal pH of 6.40.

Our data (Fig. 2) allow an approach for calculating the permeabilities of \( \text{NH}_3 \) and \( \text{NH}_4^+ \). The slope of the regression should reflect the impact of the increase in \( \text{NH}_3 \), and its numerical value represents the flux rate of \( \text{NH}_3 \) per millimole per liter of \( \text{NH}_3 \) applied. The intercept of ammonia flux on the y-axis (0.7 \( \mu \text{eq} \cdot \text{cm}^{-2} \cdot \text{h}^{-1} \)) predominantly or solely represents \( \text{NH}_4^+ \) flux at a luminal concentration of 30 mM, from which the flux rate of \( \text{NH}_4^+ \) per millimole per liter of \( \text{NH}_3 \) can be derived.
by simple division. The permeability ratio $P_{NH_3}/P_{NH_4}$ can be obtained by dividing the slope of the regression by this value, yielding a value of $P_{NH_3}/P_{NH_4} = 175$. The value that is reported for the human colon ($400$) is even higher ($15$). Thus the effects of luminal pH on the absorption of ammonia, and Na transport via the NHE, should be considerable in both tissues.

Despite the high value of $P_{NH_3}/P_{NH_4}$, which can lead to the incorrect assumption that transport of $NH_4^+$ is "negligible," the amount of ammonia absorbed in the ionic form is considerable due to the low concentration of $NH_3$ in relationship to $NH_4^+$ at physiological levels of pH. At neutral pH, and a total concentration of 30 mM ammonia, only 0.27 mM is in the form of $NH_3$ so that even if the permeability of $NH_4^+$ is 175 times lower than that of $NH_3$, the concentration is higher by a factor of 110. This means that for every millimole per liter crossing the membrane as $NH_3$, only $0.27 \times 110 = 29.7$ millimoles per liter cross as $NH_4^+$. At a slightly acidic pH of 6.4 ($[NH_3] = 0.07$ mM), conversely, 70% is absorbed as $NH_4^+$, whereas at a pH of 7.4 ($[NH_3] = 0.68$ mM), 80% of ammonia is absorbed as $NH_3$. This is in good agreement with our observations concerning the stimulation of sodium transport via the NHE at a pH of 6.4 and its inhibition at a pH of 7.4.

The patch-clamp data in this study show that ruminal epithelial cells express quinidine-sensitive potassium channels, as suggested previously (5). Whereas additional uptake of ammonia through specific transporters cannot be ruled out (31, 53, 54), these transporters cannot explain the large $NH_4^+$-induced currents observed in both the Ussing chamber and patch-clamp data in this study. Note that transporters from the Rh gene family are typically saturated by a few millimoles per liter of $NH_4^+$, and that transport is not affected by a change in membrane potential (54).

Conductance of $NH_4^+$ through nonselective cation channels has been reported for a number of other tissues (10). In rumen, this appears unlikely. The internal solution for patch-clamp experiments was chosen to minimize contributions of the nonselective cation channel (36), and no appreciable increase in conductance of sodium compared with choline could be observed. Note that under physiological conditions with divalent containing solution, the sodium conductance of rumen epithelium is saturated at 30 mM Na$^+$ (20, 47). However, exposure to $NH_4^+$ at pH 6.4 induced an additive increase in the transport rates of $NH_4^+$ and Na$^+$, with a corresponding increase in $I_{Na}$. Thus there was no sign of competition between Na$^+$ and $NH_4^+$ for a common pathway. In contrast, K$^+$ and $NH_4^+$ interfered with each other both in the patch-clamp and Ussing chamber experiments.

The question may arise if the currents observed in the patch-clamp measurements were induced by cell swelling or changes in pH due to influx of $NH_3$. However, capacitance
remained stable, and both induction of current and depolarization could be observed at values of pH (5.8, 6.4), at which the concentration of NH₃ was negligible. Note also that stimulation of current was observed in all cases. Stimulation of potassium and/or chloride currents should induce hyperpolarization in K-glucuronate-filled cells, and not the depolarization observed. Effects of ammonia persisted after the removal of either intracellular chloride (as in the K-glucuronate experiments) or potassium (as when NH₄ or choline was used to replace K in the pipette solution). The experiments also show that reversal potential increases when NH₄Cl is used to replace choline chloride externally and decreases when the same is done internally, in line with our assumption.

The NH₄Cl-induced inward current was blocked by BaCl₂ or quinidine but not by iberiotoxin. Single-cell experiments showed a conductance of ~130 pS in symmetrical NH₄Cl solution that was permeable to both K⁺ and NH₄⁺ with a P_NH₄/P_K of 0.5 ± 0.5. This permeability ratio, derived from the reversal potential (29), corresponds roughly to the slightly higher value of P_NH₄/P_K = 0.7 ± 0.4 found in whole cell experiments at a 130 mM concentration of both ions.

However, note that under conditions in which the pore is saturated, the permeability ratio as derived from the reversal potential deviates from the ratio of the absolute permeabilities, as derived by dividing the absolute current values (29). This effect can be seen clearly in Fig. 4C, where current is not doubled by doubling the concentration of NH₄⁺, as independence theory would demand.

Permeation properties of K⁺ channels are generally similar in that they must reconcile two apparently contradictory properties: a high rate of ion conductance and a high selectivity (4, 29). A single-file, multi-ion pore is the preferred model and predicts the effects that we observed in ruminal epithelial cells, such as the fact that currents induced by a mix of K⁺ and NH₄⁺ were lower than the sum of the currents induced by each ion alone (Fig. 4, C and D); in other words, that the two ions interfered with each other when flowing through the lumen of the pore(s). Typically, effects on reversal potential were not pronounced: at the reversal potential, the flux of ions through the pore comes to a halt.

Another seemingly paradox observation is that in the current study, we observed stimulation not only of inward, but also of outward current when cells were exposed to either K⁺ or NH₄⁺ from the external side.

Again, the multi-ion channel model predicts such effects. Electrostatic repulsion speeds ion flow when such channels become multiply occupied. In many cases, channel pores have to be fully occupied (with a distinct number of ions) before a current begins to flow. Another property of multi-ion channels leading to concentration-dependent permeability is that permeant ions entering the pore from one side may nudge out ions blocking the pore from the inside. Thus the channel can discriminate between currents into the cell and out of it; in other words: the channel is (inwardly or outwardly) rectifying (29).

Alternately, occupancy of an ion-selective site in the pore by a permeant ion from either the outside or the inside via the “foot-in-the-door” mechanism is known to prevent entry into the inactivated state in potassium channels with C-type inactivation (29, 56).

These basic properties of potassium channels may explain an old paradox. It is undisputed that elevation of potassium depolarizes the membrane of ruminal epithelium (20, 35, 40) and that potassium is absorbed across the ruminal wall (50),
stimulating $I_{ec}$ (5). However, various studies have shown that in NaCl Ringer, apical secretion of potassium is minimal (21, 35) despite the activity of the basolateral Na/K-ATPase.

We suggest that permeability for potassium is, indeed, low in apical NaCl solution and may be further reduced by the presence of $NH_4^+$ but is stimulated by the elevation of mucosal potassium. The physiological importance of this mechanism is evident in that it prevents a drain of potassium from the serosal side into the ruminal lumen under low-potassium conditions. Note that in multi-ion channel theory, $NH_4^+$ ions should be able to “sweep” individual potassium ions caught in the single file of $NH_4^+$ streaming through the pore into the cell, even against an electrochemical gradient for potassium (29).

Beyond this, we are unable to precisely identify the potassium channel or, more likely, group of channels that conducts ammonia in ruminal epithelium. Unfortunately, data on the large family of potassium channels are incomplete (27) and depend greatly on the tissue in which the channel is expressed. Conductance of $NH_4^+$ by various potassium channels is well documented (27), as is the presence of potassium channels in the apical membrane of intestinal tissues (18), which may represent a pathway for $NH_4^+$ efflux (8) from the lumen. Large-conductance calcium-activated potassium channels have been reported to conduct $NH_4^+$, but these channels display only minimal open probability at hyperpolarized voltages, ruling out these channels as a possible route for $NH_4^+$ influx in our experiments, as does the lack of an effect of iberiotoxin.

It is difficult to resist speculation concerning the role of $NH_4^+$ for the pathophysiologic of hypomagnesaemia in ruminants. Hypomagnesaemia (grass tetany) is linked to use of artificial fertilization techniques with high concentrations of ammonia. Inhibition of magnesium absorption not only by K+ (11, 35, 36, 48) but also by ammonia has been known for many years (11, 23) as has the reduction of magnesium uptake via depolarization of the apical membrane (35, 48). Thus we suggest that uptake of $NH_4^+$ inhibits magnesium absorption via depolarization of the apical membrane, as previously described for K+ (11, 35, 48).

Care et al. (11) also noted that inhibitory effects of $NH_4^+$ and K+ on Mg$^{2+}$ uptake were additive, which is in line with additive effects of $NH_4^+$ and K+ on membrane potential as suggested by this study. Inhibition of Na transport by ammonia at a pH of ~7 was also reported, which can be attributed to the fact that at this pH, transport of $NH_3$ is larger than transport of $NH_4^+$. However, the finding that addition of K+ results in a further decrease in Na absorption is in contrast to the effects of K+ alone (34, 36, 50). We suggest that in Care’s experiments, K+ inhibits flux of $NH_4^+$ into the cells. Thus the stimulatory effect of $NH_4^+$ on NHE is terminated, and the inhibitory effect of NH$_3$ alone leads to a net inhibition of sodium transport by elevation of K+.

In conclusion, this study presents evidence that in the rumen, electroneutral Na transport is inhibited by ammonia entering the cytosol in the form of $NH_3$ and is stimulated by entry of $NH_4^+$ through potassium channels. The luminal pH determines the predominant form of luminal ammonia uptake, $NH_3$ or $NH_4^+$ and, hence, the effect of ammonia on Na transport. Because the physiologic pH of the ruminal fluid is <6.9, ammonia enhances Na absorption. This modulation of NHE activity by ammonia appears to be as important as the “classic” effect of short-chain fatty acids (24, 33). Furthermore, it is interesting to note that electronegative uptake of ammonium increases with adaptation to a high-protein diet (1).

The basic physiological mechanism outlined in this study should be valid not only for the rumen, but also for the human colon due to the similarities between the two tissues, namely electroneutral Na transport via NHE (19, 37, 43), abundance of K channels (33, 51), and a wide variation of ammonia concentrations and pH in the ingesta (9).

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REFERENCES


AJP-Gastrointest Liver Physiol • VOL 289 • SEPTEMBER 2005 • www.ajpgi.org
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Hrnjez BJ, Song JC, Prasad M, Mayol JM, and Matthews JB.


